REPORT OF

ANALYTICAL EVALUATION PROGRAM STANDARD WATER SAMPLES NUMBERS 3 and 4

Comparison of results obtained using SDA 3A with results using tax-free ethanol

> Total hardness, Sulfate, Chromium, Copper, Nickel

> U. S. GEOLOGICAL SURVEY WATER RESOURCES DIVISION Quality of Water Branch Denver, Colorado 1963

CONTENTS

Page

Purpose and plan	ŀ	•	•	ŗ			i.	÷	•	з¢	3
Preparation of the samples		ļ	•			k,	•	•	į.	•	4
Preparation of denatured alcohol		X			ś	r.	i.	٠	, . ,	÷	5
Participating laboratories							i.	•	e,	\mathbf{a}	6
Reported results: total hardness	5	i.	•	•		s,	÷	•			7
Reported results: sulfate	•	•		٩,	×,	·	h.	ŝ	•	•	1 4
Reported results: chromium		•		•	÷	e.	ķ		i.		21
Reported results: copper			÷	ł			•	۱.	÷	Ę.	28
Conclusions and recommendations											35

Tax-free ethanol is used in many analytical procedures; however, there are certain mandatory controls on its use which are both time consuming and costly. It has been proposed that, if possible, some non-controlled reagent be substituted for tax-free ethanol. Denatured alcohol is not a controlled item; and Specially Denatured Alcohol No. 3A (SDA 3A) would appear to be a satisfactory substitute because of its formulation (5 gallons commercially pure methyl alcohol to 100 gallons 190-proof ethyl alcohol) and its low cost. It may be purchased in either 5-gallon containers (\$5.95) or in 55•gallon drums (\$43.45). These prices are approximately the same as for tax-free ethanol. A permit is required for the purchase of SDA 3A, and the same procedure for ordering taxfree ethanol must be used for ordering SDA 3A. The Internal Revenue Code governing the use of specially denatured spirits by a government agency does not require that a record of use be kept (Part 211 of Title 26, Code of Federal Regulations, Subpart L, paragraphs 211.231-211.237).

There are presently four analytical procedures in Water-Supply Paper 1454 which specify the use of 95 percent ethanol. Of these, the most frequently run determinations are (1) total hardness, and (2) sulfate; the other two are chromium and copper. A nickel procedure which was distributed to District laboratories about two years ago also specifies the use of 95 percent ethanol.

In order to evaluate the feasibility of substituting SDA 3A, Specially Denatured Alcohol, for 95 percent ethanol where specified in the analytical procedures for determining total hardness, sulfate, chromium, copper, and nickel, all QW District laboratories were invited to participate in the analysis of standard reference samples, using both alcohols for comparison. The evaluation, while providing a basis for comparison of results obtained using denatured alcohol and undenatured alcohol, the program also provided a simultaneous evaluation of the methods and the laboratories performing the analyses. This report is a summary of the data submitted by the participating laboratories.

<u>1</u>/ Supplier: U. S. Industrial Chemicals, 624 South Michigan Avenue, Chicago 5, Illinois.

PREPARATION OF THE SAMPLES

Each sample was prepared from accurately weighed amounts of analytical reagent-grade chemicals dissolved in an accurately measured volume of distilled water which had been further purified by passage through a mixed-bed exchanger. When necessary to affect solution of the reagent, a slight excess of reagent-quality nitric acid was added to the sample.

The following compounds were used for the preparation of the samples:

$MgSO_4 \cdot 7H_2 O$	$NiCl_2 \cdot 6H_2 O$
K _g CrO ₄	Cu (metal)

Concentrated stock solutions were prepared to contain the following concentrations of the substances indicated:

Stock solution 3

Total hardness (as CaCO ₃)992	ppm
Sulfate (SO ₄)953	ppm
Chromium (Cr) 5.8	ppm
Copper (Cu)	ppm
Nickel (Ni) 1.8	ppm

Stock solution 4

Total hardness (as CaCO ₃)4438	ppm
Sulfate (SO₄)4265	ppm
Chromium (Cr) 32.8	ppm
Copper (Cu) 29.5	ppm
Nickel (Ni) 7.8	ppm

Individual 1-liter samples of Standard Water Sample No.3 were prepared by taking 25.0 ml of Stock solution 3 and diluting to exactly one liter. Standard Water Sample No. 4 was prepared in a similar way by diluting a 25.0-ml portion of Stock solution 4 to exactly one liter. The calculated concentrations of the two samples thus prepared were as follows:

	Standard Water	Sample
	No. 3	No. 4
Total hardness (as CaCO ₃)	25	111
Sulfate (SO ₄)	24	107
Chromium (Cr)	0.14	0.82
Copper (Cu)	0.17	0.74
Nickel (Ni)	0.04	0.20

After preparation, each sample was analyzed in duplicate using both tax-free ethanol and SDA 3A at two different times over a period of about four weeks. The results of these analyses are given in the following table. Neither sample showed any change in the concentration of the substances determined over the period of storage.

	Standard Water Sample							
	No	• 3	No	. 4				
	Tax-free		Tax-free					
	ethanol	SDA 3A	ethano1	SDA 3A				
Total hardness(as CaCO ₃)	24 ppm	24 ppm	110 ppm	110 ppm				
Sulfate (SO ₄)	25	25	108	107				
Chromium (Cr)	0.13	0.14	0.80	0.79				
Copper (Cu)	0.19	0.20	0.78	0.79				
Nickel (Ni)	0.05	0.04	0.18	0.18				

Analysis by Preparations Laba/

 \underline{a} / Each result represents the average of 2 duplicate determinations made over a period of about 4 weeks.

PREPARATION OF DENATURED ALCOHOL

Specially denatured alcohol No. 3A was prepared by adding one liter of commercially pure methyl alcohol to 20 liters of 190-proof ethyl alcohol. One liter of SDA 3A was sent to each of the participating laboratories.

PARTICIPATING LABORATORIES

Alaska, Palmer California, Menlo Park California, Sacramento Colorado, Denver Florida, Ocala Louisiana, Baton Rouge Nebraska, Lincoln New Mexico, Albuquerque

New York, Albany North Carolina, Raleigh Ohio, Columbus Oregon, Portland Pennsylvania, Philadelphia Texas, Austin Utah, Salt Lake City Wyoming, Worland





REPORTED RESULTS: TOTAL HARDNESS (ppm)

Code	Tax	-free etha	nol	SDA 3A					
No.	(1)	(2)	Avge.	(1)	(2)	Avge.			
101	25	26	26	26	26	26			
102	5.9	6.1	6.0 <u>a</u> /	6.3	6.1	6.2 ª/			
103	25	26	26	26	26	26			
104	25	25	25	25	25	25			
105	23	24	24	24	24	24			
1 06	25	25	25	25	25	25			
107			<u></u> - ,						
108	25.0	25.0	25.0	24.0	25.0	24.5			
109	26	26	26	26	26	26			
110	25	25	25	25	25	25			
111	25	25	25	25	25	25			
112	26 26	27 27	26	26 26	27 26	26			
113	27	28	27 <u>b</u> /	28	27	27 <u></u> ^b /			
114	26	26	26	26	26	26			
115	25.5	26.0	26	25.5	26.0	26			
116	25	25	25	25	25	25			

Standard Sample No. 3

<u>a</u>/ Reported as magnesium; ≅ 25 ppm T.H. and ≅ 26 ppm T.H., respectively.

b/ Should have been reported as 28 ppm.

REPORTED RESULTS: TOTAL HARDNESS (ppm)

Code	Tax	-free eth	anol	SDA 3A					
No.	(1)	(2)	Avge.	(1)	(2)	Avge.			
101	116	116	116	114	116	115			
102	27	27	27 ª/	28	28	28 <u>a</u> /			
103	123	121	122	123	123	123			
104	112	112	112	112	112	112			
105	108	108	108	109	109	109			
106	110	110	110	110	110	110			
107				()	e Perios				
108	112	113	112.5	113	112	112.5			
109	113	114	114	114	115	114			
110	108	107	108	108	108	108			
111	114	114	114	114	114	114			
112	114 117	114 114	115	116 114	114 114	114			
113	113	114	113 <u>b</u> /	115	115	115			
11 4	113	114	114	115	114	114			
115	113	114	114	113	114	114			
11 6	113	113	113	113	113	113			
						1			

Standard Sample No. 4

<u>a</u>/ Reported as magnesium; ² 111 ppm and 115 ppm T.H., respectively.

b/ Should have been reported as 114 ppm.





Methods used: Hardness (as CaCO₃)

Lab.	Me	thod	Modifications
101	WSP 1454,	D:17a-1	Porcelain evaporating dishes; no blank correction; glycerin solvent for EBT indicator.
102	WSP 1454,	D:23a-3	Magnesium determined.
103	WSP 1454,	D:17a-1	None
1 04	п	n	"
105	u	. H	No. 2 evaporating dishes used.
106		н	None
107	(not deter	rmined)	
108	WSP 1454,	D:17a-1	병두 전 그는 것이 가격하는 것을 해야 했다.
109	(not desig	gnated)	
110	WSP 1454,	D:17a-1	None
111		"	NH ₂ OH·HC1, NH ₄ OH added in order (1.5 ml each) with dropping pipet; 10-ml buret used for hardness titration.
112	"		물로 그 가 모두 한 것을 하는 것 같아?
113		"	Tax-free alcohol-EBT indicator was several months old.
114			NaCN added before $NH_4 OH$.
115	п	n	None
116	U.	"	

HARDNESS DETERMINATION

Standard Water Sample No. 3, 25 ppm

Errors

Error (absolute)		Number laboratories	Number of laboratories reporting			Percentage of 15 laboratories reporting					
		Tax-free ethanol	SDA 3A	Tae	ax-free thanol	SDA 3A					
0	ppm	7	5	47	percent	33	percent				
±1	11	14	14	93	0	93	11				
±2	11	15	15	100	- 11	100	"				

Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol

Laboratories	reporting	-1	ppm	difference	- 1		
	. ч	0			13	(87	percent)
가지는 병원을 한	н. Н	+1	11		1		

Standard Water Sample No. 4, 111 ppm

Errors

Error (absolute)		Number of laboratories reporting		Percentage of 15 laboratories reporting						
		Tax-free ethanol	SDA 3A	Ta	ax-free thanol	SDA 3A				
0	ppm	1	0	7	percent	0	percent			
±1		4	3	27		20				
±2	н	6	5	40	н	33				
±3		12	11	80		73	н			
±4	н	13	14	87		93				
±5		14		93	н					
±11	n (1	15		100	п					
±1 2			15			100	п			

Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol

Laboratories	reporting	-1	ppm	difference	-	2		
	11	0	11			9	(60	percent)
11	п	+1	11	**		2	-	
11	- - P	+2	.11	11		1		
п		+3	"			0		
	, 'u - jii	+4		"		1		

The evaluation of the data for total hardness shows no significant difference in results if denatured alcohol (SDA 3A) is substituted for tax-free ethanol. Eighty-seven percent of the laboratories reported identical values for Sample No. 3 and 60 percent reported identical values for Sample No. 4. The remaining laboratories, however, reported both higher and lower values using denatured alcohol. Nevertheless, all results are within the limits of the method.

The most probable values for hardness in Samples Nos. 3 and 4 are 25 ppm and 111 ppm, respectively. There were as many laboratories reporting 26 ppm hardness in Sample No. 3 as those reporting 25 ppm. This again indicates, as was shown in a previous study (Standard Water Samples Nos. 1 and 2), a bias in favor of reporting even-numbered values. Most of the values reported for Sample No. 4 are higher than the calculated value. Perhaps a blank correction is being neglected. The data for Sample No. 4 also indicates that reporting the results to 1 ppm probably is not justified.

REPORTED RESULTS: SULFATE (ppm)

Code	Tax	x-free eth	ano1	SDA 3A				
No.	(1)	(2)	Avge.	(1)	(2)	Avge.		
101	22	23	22	23	22	22		
102	24	24	24	24	24	24		
103	25	24	24	25	23	24		
104	29	29	29	29	29	29		
105	23	24	24	23	24	24		
106	24	24	24	24	24	24		
107	26	24	25	24	25	24		
108	24.0	24.0	24.0	25.0	<u>a</u> /	25.0		
109	24	24	24	24	24	24		
110	24	24	24	24	22	23		
111	23	24	24	24	24	24		
112	24 23	24 23	24	24 23	25 24	24		
113	27	26	26	25	25	25		
11 4	24	24	24	24	24	24		
115	22	23	22	22	23	22		
116	24	23	24	24	23	24		
					and the second se			

Standard Sample No. 3

<u>a</u>/ Omitted; bad indicator solution.

REPORTED RESULTS: SULFATE (ppm)

Code	Tax	-free eth	ano1	SDA 3A				
No.	(1)	(2)	Avge.	(1)	(2)	Avge.		
101	103	102	102	102	102	102		
102	106	105	106	1 04	1 04	104		
103	118	119	118	117	119	118		
104	101	101	101	101	101	101		
105	99	99	99	99	99	99		
106	108	108	108	108	108	108		
107	105	106	106	104	1 04	104		
108	104	104	104	108	107	107.5		
109	109	109	109	109	109	109		
110	108	109	108	105	107	106		
111	108	108	108	108	108	108		
112	106 107	107 106	106	107 107	108 106	107		
113	106	106	106	106	106	106		
114	108	108	108	108	108	108		
115	105	105	105	105	105	105		
116	106	106	106	106	106	106		
Acres 1 and	index of the last	A second s		I a second second second		Construction and the second s second second sec		

Standard Sample No. 4



SULFATE STANDARD SAMPLE NO. 4



Lab.	Method	Modifications
101	Spectrophotometri thorin. WSP 1454 D:38a-2	Lumetron colorimeter; K=490 mµ. E.P. at 0.19 absorbance.
102	H	None
103	п п	· · ·
104	п п	"
105	о п	Titration from absorbance of 0.100 to 0.300; 100-ml beakers used in- stead of 50-mm cells.
1 06	и и	None
107	n n	Titration from absorbance of 0.100 to 0.300.
108		None
109		П
110	п п	TI
111	Visual thorin. WSP 1454, D:38a-1	ETOH used for thorin reagent; ex- change columns are 18"x1"-diameter tubes containing a 10"-column of Amberlite resin; flow rate of
		sample is approx. 20 ml/min.
112	"	- North Anna 2017년 1월
113	(not designated)	
114	Spectrophotometri thorin. WSP 145 ¹ D:38a-2	ic Solvent-indicator solution unstable therefore separate solutions of each prepared; indicator 0.5 g thorin and 10 g NaOAc per 500 ml; solvent 12 ml HOAc per 1,000 ml alcohol; 1 ml of indicator and 40 ml of solvent added to each sample.
115	н. 1. П. 1. ст. т	None
116	U U	pH of sample adjusted to 2.5 be- fore indicator added.

Methods used: Sulfate (SO₄)

SULFATE DETERMINATION

Standard Water Sample No. 3, 24 ppm

Error (absolute)		Numbo laboratorio	er of es reporting	Percentage of 16 laboratories reporting					
		Tax-free ethanol	SDA 3A	Tax-free ethanol	S	SDA 3A			
0	ppm	11	10	69 percen	t 62	percent			
±1		12	13	75 "	81	11			
±2	11	15	15	94 "	94	n			
±5	11	16	16	100 "	100				

Errors

Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol

Laboratories	reporting	-1	ppm	difference	- 3		
11		0	11		12	(75	percent)
11		+1		n	1		

Standard Water Sample No. 4, 107 ppm

Errors

Error (absolute)		Numbe laboratorie	r of s reporting	F labo	Percentage of 16 laboratories reporting						
		Tax-free ethanol	SDA 3A	Tax eth	anol	SD	SDA 3A				
0	ppm	0	1	0 p	ercent	6 p	ercent				
±1	11	9	8	56	n -	50	11				
±2	11	11	10	69		62					
±3		12	12	75	-"	75	11				
±5		13	13	81	н	81					
±6	11	14	14	87	"	87	п				
±8		15	15	94		94	"				
±11	0	16	16	100		100					



Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol

Laboratories	reporting	-2	ppm	difference	- 3	
11		-1		U I	0	
u.	- H 12	0	11		11	(69 percent)
11		+1		U.	1	전 김 씨는 영화
11		+4	11	11	1	

The calculated values for sulfate were 24 ppm (Sample No. 3) and 107 ppm (Sample No. 4). The bias in favor of reporting even-numbered values can easily be seen from this data. Eleven of the 16 laboratories reported 24 ppm for Sample No. 3 and, of these, 9 also reported perfect results using denatured ethanol. For Sample No. 4 only one laboratory reported a perfect result; more than half the laboratories reported either 106 ppm or 108 ppm. There was considerable spread in the results for Sample No. 4 with a tendency to report values lower than the calculated value. The results were good for Sample No. 3 with 94 percent of the laboratories reporting within ± 2 ppm of the calculated value.

The use of denatured alcohol presents no problem in the sulfate determination. More than 75 percent of the participating laboratories obtained identical values using both alcohols and for either sample. The other results were either positive or negative with a slight trend in the negative direction. REPORTED RESULTS: CHROMIUM (ppm)

Code	Tax-	free etha	inol	SDA 3A				
No.	(1)	(2)	Avge.	(1)	(2)	Avge.		
101		1						
102				n Angel		gaing tite.		
103	0.16	0.16	0.16	0.16	0.15	0.16		
104	0.15	0.13	0.14	0.13	0.14	0.14		
105	0.14	0.14	0.14	0.14	0.14	0.14		
106		15			1 			
107				1.1.1.1	e tra sé	- 		
108	0.15	0.15	0.15 ^{<u>a</u>/}	1				
109	is			. 				
110	3	[1]				이 가슴 이 같은 것		
111	0.13	0.13	0.13	0.13	0.13	0.13		
112		J 21	<u>.</u>					
113								
11 4				i , i				
115		1		1 <u></u> 1 - 1				
116	0.07	0.07	0.07 <u>b</u> /	0.08	0.08	0.08 <u>b</u> /		

Standard Sample No. 3

a/ 1:1 (acetone:water) substituted for ethanol.

b/ Calculation error; later corrected by participating laboratory to 0.14 ppm, tax-free ethanol, and 0.16 ppm, SDA 3A.



REPORTED RESULTS: CHROMIUM (ppm)

Code	Tax-	free etha	nol	SDA 3A				
No.	(1)	(2)	Avge.	(1)	(2)	Avge.		
101								
102						동생 같아.		
103	0.80	0.80	0.80	0.80	0.81	0.80		
1 04	0.84	0.82	0.83	0.83	0.83	0.83		
105	0.78	0.78	0.78	0.78	0.78	0.78		
1 06								
107					- 111			
108	0.87	0.87	0.87 <u>a</u> /					
109			, - i , , i	"	0.5 ³ - 1			
110			+-	·		i se nijela		
111	0.79	0.79	0.79	0.79	0.78	0.78		
112		학~~ 영상	14 - Jan 19	; i		9 N.S		
113			-4.1		-7			
114	() · · · · · · · · · · · · · · · · · ·	in second						
115					1 4 1			
116	0.37	0.36	0.36 <u>b</u> /	0.36	0.39	0.37 <u></u> ^b /		

Standard Sample No. 4

a/ 1:1 (acetone:water) substituted for ethanol.

b/ Calculation error; later corrected by participating laboratory to 0.72 ppm, tax-free ethanol, and 0.79 ppm, SDA 3A.



STANDARD SAMPLE NO.4



a/ 1:1 (acetone:water) substituted for ethanol

Lab.	Method	Modifications
101	(not determined)	8 이 이 방법 전 이 이 방법 수영 같이.
102	и и	그 그 그 가 나 가 가 가 가 다 가 가 다 가 다 가 다 가 다 가 다 가 다
103	WSP 1454, D-12a-1	None
104	<u>и</u> и	
105	, п п	
106	(not determined)	
107	н	
108	Diphenylcarbazide	Diphenylcarbazide reagent prepared in 1:1 acetone:water.
109	(not determined)	
110	п. п.	
111	WSP 1454, D:12a-1	None
112	(not determined)	
113		
114	н	
115	н	
11 6	WSP 1454, D:12b-1	None

Methods used: Chromium (Cr)

CHROMIUM DETERMINATION

Standard Water Sample No. 3, 0.14 ppm

Errors

Error (absolute)		Number of <u>laboratories reporting</u> Tax-free						<u>lal</u> Ta	Percentag <u>laboratories</u> Tax-free			e of 4 <u>reporting</u> #/		
		ethanol		SDA 3A		e	thano1		SDA 3A					
0.00	ppm		2				2	50	percent	50	pe	rcent		
±0.01	11		3				3	75	.11	75		11		
±0.02	11		4				4	100	11	100	es [†]	11		

a/ One laboratory reported 0.15 ppm; they substituted 1:1 (acetone:water) for ethanol. Another laboratory reported 0.07 ppm (using tax-free ethanol) and 0.08 ppm (using SDA 3A), later changed to 0.14 ppm and 0.16 ppm, respectively, after correcting for a calculation error. These results are not included in the evaluation.

Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol

Laboratories reporting 0.00 ppm difference - 4 (100 percent)

Standard Water Sample No. 4, 0.82 ppm

Errors

Error (absolute)		Number of laboratories reporting			Percentage of 4 laboratories reporting ^a				
		Tax-free ethanol	SDA 3A	et	hanol	SDA 3A			
0.00 pp	m	0	0	0	percent	0	percent		
±0.01 "	t	1	1	25		25	- 11		
±0.02 "	1	2	2	50	_11 -	50	11		
±0.03 "	е÷.,	3	2	75	"	50			
±0.04 "		4	4	100	11	100	11		

<u>a</u>/ One laboratory reported 0.87 ppm; they substituted 1:1 (acetone:water) for ethanol. Another laboratory reported 0.36 ppm (using tax-free ethanol) and 0.37 ppm (using SDA 3A), later changed to 0.72 ppm and 0.74 ppm, respectively, after correcting for a calculation error. These results are not included in the evaluation.



Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol

Laboratories reporting -0.01 ppm difference - 1 " 0.00 " " 3 (75 percent)

Only four laboratories participated in the chromium determination. The results were excellent. For Sample No. 3, the four participating laboratories were within ± 0.02 ppm of the calculated value (0.14 ppm), and for Sample No. 4 (0.82 ppm) were within ± 0.04 ppm.

The choice of solvent for the diphenylcarbazide reagent does not affect the results. One laboratory used neither ethanol nor denatured alcohol, but used a 1:1 ratio of acetone and water, and their results were as good as the other data submitted using denatured or tax-free ethanol.

REPORTED RESULTS: COPPER (ppm)

Code	Tax	-free eth	anol	SDA 3A				
No.	(1)	(2)	Avge.	(1)	(2)	Avge.		
101	0.20	0.20	0.20	0.20	0.20	0.20		
102								
103	0.15	0.15	0.15	0.15	0.15	0.15		
1 04	0.18	0.18	0.18	0.19	0.19	0.19		
105	0.17	0.17	0.17	0.15	0. 1 5	0.15		
106			<u>-</u>	· · ·	1			
107	0.20	0.20	0.20	0.20	0.20	0.20		
108	0.19	0.19	0.19 ^a /	11 1 -011	State State			
109				- <u>-</u>	1			
110					승수 문			
111	0.16	0.16	0.16	0.15	0 .1 5	0.15		
112	0.15		0.15	0.15	금부 문	0.15		
113	0.08	0.07	0.08 <u>b</u> /	0.07	0.07	0.07 <u>b</u> /		
114	0.17	0.17	0.17	0.18	0.18	0.18		
115		a '						
116	0.17	0.17	0.17	0.17	0.17	0.17		
and the second se			4. No. 1 (1997)					

Standard Sample No. 3

a/ Methanol substituted for ethanol.

b/ Calculation error; later reported by participating laboratory as 0.19 ppm (tax-free ethanol) and 0.18 ppm (SDA 3A).

REPORTED RESULTS: COPPER (ppm)

Code	Tax	free eth	anol		SDA 3A				
No.	(1)	(2)	Avge.	(1)	(2)	Avge.			
101	0.75	0.75	0.75	0.75	0.75	0.75			
102		, '''			Bank P				
103	0.75	0.75	0.75	0.75	0.75	0.75			
104	0.86	0.85	0.86	0.87	0.88	0.88			
105	0.75	0.75	0.75	0.75	0.74	0.75 ^a /			
106		da 1 (2)	그는 일을						
107	0.80	0.84	0.82	0.80	0.84	0.82			
108	0.76	0.76	0.76 <u>b</u> /		%				
109									
110		티 이분			12				
111	0.78	0.78	0.78	0.78	0.78	0.78			
112	0.77		0.77	0.75		0.75			
113	0.31	0.31	0.31 ^c /	0.31	0.31	0.31 ^{c/}			
114	0.77	0.77	0.77	0.77	0.78	0.78			
115		s	+-						
116	0.77	0.77	0.77	0.81	0.80	0.80			
		and the second sec		11	the second se				

Standard Sample No. 4

a/ Should have been reported as 0.74 ppm.

b/ Methanol substituted for ethanol.

<u>c</u>/ Calculation error; later reported by participating laboratory as 0.77 ppm (tax-free ethanol) and 0.77 ppm (SDA 3A).



COPPER STANDARD SAMPLE NO. 3



COPPER STANDARD SAMPLE NO. 4



a / Methanol substituted for ethanol

Lab.	Method	Modifications
101	WSP 1454, D:14a-1	None
102	(not determined)	
103	WSP 1454, D:14a-1	None
104	— п п	н
105	п	50-mm optical depth cells used.
106	(not determined)	
107	WSP 1454, D:14a-1	None
108	Carbamate method	MeOH substituted for ETOH.
109	(not determined)	
110	11 11	
111	WSP 1454, D:14a-1	None
112	н	
113	пп	"
114	п п	"
115	(not determined)	
116	WSP 1454, D:14a-1	None

Methods used: Copper (Cu)

COPPER DETERMINATION

Standard Water Sample No. 3, 0.17 ppm

Errors

Error (absolute)		Number of <u>laboratories reporting</u>			Percentage of 9 laboratories reporting ^a /				
		Tax-free ethanol	SDA 3A	Ta et	ax-free thanol	SDA 3A			
0.00	ppm	3	1	33	percent	11	percent		
±0.01	н	5	2	55	н	22	11		
±0.02	11	7	7	78	. Û	78	"		
±0.03	11	9	9	100	U I	100	11		

<u>a</u>/ One laboratory reported 0.19 ppm; they substituted methanol for ethanol. Another laboratory reported 0.08 ppm (using tax-free ethanol) and 0.07 ppm (using SDA 3A), later changed to 0.19 ppm and 0.18 ppm, respectively, after correcting for a calculation error. These results are not included in the evaluation.

Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol

Laboratories	reporting	-0.02	ppm	difference	-	1			
11		-0.01	Ú.	с н		1			
11		0.00	11	н		5	(55	percent)	
н	11	+0.01	11			2			

Standard Water Sample No. 4, 0.74 ppm

Error (absolute)		Number of <u>laboratories reporting</u> Tax-free ethanol SDA 3A			labo Taz	$\frac{a}{1}$			
0.00 p	Dm	0	<u> </u>	0		0 1	percent	<u></u>	percent
±0.01	11	3		4		33		44	11
±0.03	u.	6				67	"		<u>_</u> 64
±0.04	n	7		6		78	"	67	31
±0.06	11	1 1 1		7				78	
±0.08	81	8		8		89	п	89	n -
±0.12	11	9				¦ 100			
±0.14	11			9				100	п

Errors

<u>a</u>/ One laboratory reported 0.76 ppm; they substituted methanol for ethanol. Another laboratory reported 0.31 ppm (for both alcohols), later changed to 0.77 ppm (for both alcohols) after correcting for a calculation error. These results are not included in the evaluation.

Comparison of results obtained using SDA 3A with results obtained using tax-free ethanol Laboratories reporting -0.02 ppm difference - 1 -0.01 5 (55 percent) 0.00 +0.01 +0.02 +0.03

Eleven laboratories determined copper. However, only 9 sets of data were used in the evaluation since one of the two laboratories not included used methanol in place of ethanol, while the other had calculation errors. The copper method appears to be satisfactory, with 78 percent of the data falling within ± 0.02 ppm of the calculated value for Sample No 3, and between ± 0.04 and ± 0.06 ppm for Sample No. 4.

SDA 3A presents no problem in the copper determination; more than half of the laboratories reported identical results using either alcohol. The remaining laboratories reported results on both sides of their tax-free alcohol value. These differences were within the limits of the method.

CONCLUSIONS AND RECOMMENDATIONS

From the data for each method tested, denatured alcohol (SDA 3A) can be substituted for tax-free ethanol with no adverse effect. The data from the sulfate determination were superior to the data from the other methods, and the sulfate method gives us the best test for this study since 80 percent of the sample solution is alcohol and 5 percent of this is methanol. In the other methods, only minute amounts of methanol from the SDA 3A formulation are present.

Good results were obtained by one laboratory using methanol in the copper procedure and a 1:1 ratio of acetone to water in the chromium procedure. It is possible that these solvents may be used as substitutes for ethanol in these two determinations.

Hardness

1. Hardness concentrations of the order of 25 ppm can be determined to within ± 1 ppm.

2. Hardness concentrations above 100 ppm cannot be determined to within ± 1 ppm by the present method. The reliability of the present method for determining hardness above 100 ppm is approximately ± 3 ppm, and results should be reported with this notation.

Sulfate

1. Concentrations of the order of 24 ppm can be reported accurately to within ± 1 ppm.

2. Sulfate concentrations greater than 100 ppm cannot be determined to within ±1 ppm by the present method. The accuracy of the method at this concentration level is probably ±5 ppm, and results should be reported with this notation.